

Thermodynamic Analysis of Autonomous Parallel β -Sheet Formation in Water

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Protein folding patterns are dominated by two regular substructures, α -helix and β -sheet. The origins of conformational stability for these secondary structures, a subject of intense interest, can be explored with peptides that form a helix or sheet in the absence of a tertiary context. Design rules for medium-length peptides that form autonomous α -helices in aqueous solution were delineated in the 1980s,¹ and comparable achievements for autonomous antiparallel β -sheets were reported in the 1990s.² In both cases, the development of strategies for determining folded populations has allowed thermodynamic analysis of secondary structure formation.³

An autonomously folding *parallel* β -sheet (in contrast to an autonomous antiparallel β -sheet or α -helix) cannot be created exclusively from α -amino acid residues because the N-terminus of one strand in a parallel β -sheet does not lie near the C-terminus of a neighboring strand. Many groups have explored nonpeptide units that promote parallel sheet interactions, most commonly by linking C-termini with a short turn-forming diamine.⁴ We have previously reported that the D-prolyl-1,1-dimethyl-1,2-diaminoethane (D-Pro-DADME) unit supports formation of a two-stranded parallel β -sheet in water, as indicated by 2D NMR data.⁵ Here we show that the thermodynamics of parallel β -sheet formation can be evaluated in such a model system. This accomplishment provides a foundation for exploring sequence–stability relationships in the parallel β -sheet structural manifold, which would fill a significant gap in our understanding of proteins and protein aggregates.

NMR chemical shifts provide the most reliable insight on folded populations among antiparallel β -sheet model systems,^{3,6} and we therefore aimed for a chemical shift-based analysis of parallel β -sheet folding. Most autonomous helix or sheet systems are not fully folded under accessible conditions; therefore, chemical shifts for such systems are population-weighted averages of the contributions from the limiting folded and unfolded states. To analyze **1**, we have built upon the strategy previously developed for antiparallel β -sheets by constructing model compounds intended to provide empirical estimates of chemical shifts in the fully folded and fully unfolded states.^{3b} The unfolded state is represented by diastereomer **2**; changing proline configuration from D to L completely disrupts parallel β -sheet formation.^{5,7} The folded state is represented by cyclic molecule **3**, an analogue of **2** in which the N-termini of the strand segments have been linked via a succinyl-glycine unit. Cyclization is intended to enhance the propensity for parallel β -sheet formation in **3** relative to linear molecule **2**. Among antiparallel model systems, two-stranded β -sheet conformations have been stabilized by cyclization strategies involving either the backbone (capping a β -hairpin with a β -turn) or side chains (disulfide formation between terminal Cys residues).^{3a,b,8} We focused on backbone cyclization because Cys disulfide cross-links are not compatible with parallel β -sheet secondary structure.⁹ The succinyl-glycine linker in **3** allows formation of a 10-membered-ring H-bond [C=O(Succ-18) to H–N(Ser-2)], which is analogous to the H-bond common among β -turns.

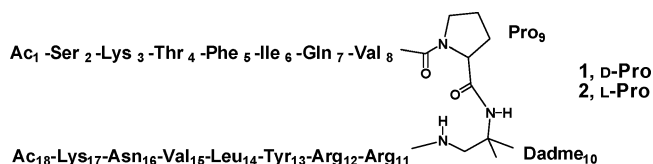


Figure 1. Chemical structures of linear compounds **1** and **2**. The numbering scheme was chosen to allow easy comparison between the linear and cyclic molecules.

NMR analysis indicated that **3** adopts the intended parallel β -sheet conformation in aqueous solution, and the data suggest that the extent of folding is greater for cyclic **3** than for linear analogue **1**. Four unambiguous interstrand NH–C α H NOEs were observed in the center of the intended β -sheet region of **3** (Figure 2); the Val-8/Arg-11 NOE, if present, would have been obscured by the residual solvent resonance. These NOEs are consistent with the expected parallel β -sheet hydrogen-bonding registry.¹⁰ The lack of interstrand NH–C α H NOEs near the succinyl-glycine linker, however, suggests that this region is not as well folded as the rest of the molecule. A large set of side chain–side chain NOEs was observed for **3**, all consistent with the intended parallel β -sheet conformation. For linear molecule **1**, a comparable set of cross-strand NOEs was seen for only the eight strand residues nearest to the D-Pro-DADME unit (Phe-5 to Val-8 and Arg-11 to Leu-14), but not for the three residues at each strand terminus.⁷ The difference in interstrand NOE patterns observed for **1** vs **3** suggests that the parallel β -sheet secondary structure encompasses nearly the entire strand length for **3**, but only the strand residues closer to the turn segment in **1**. The behavior of **1** is consistent with evidence that the termini of antiparallel β -sheet model systems tend to be unfolded (“frayed”) in aqueous solution.² As intended, macrocyclization, in **3**, strongly discourages terminal fraying.

Downfield shifts in α -proton resonances ($\delta C_{\alpha}H$), relative to a random coil reference value, indicate participation in β -sheet secondary structure.⁶ We use **2** to provide the “random coil” $\delta C_{\alpha}H$ values for assessment of folding in the strand regions of **1** and **3** because **2** shows no sign of folding (no NOEs between sequentially nonadjacent residues),⁷ and the $\delta C_{\alpha}H$ values measured for **2** account for the effects of sequence context. Nearly all strand residues in **3** show $\Delta\delta C_{\alpha}H [= \delta C_{\alpha}H(\mathbf{3}) - \delta C_{\alpha}H(\mathbf{2})] \geq +0.1$ ppm in aqueous solution at 287 K, which suggests extensive β -sheet formation along the entire length of each strand segment. This conclusion matches that reached from the interstrand NOEs observed for **3**. In contrast, the outermost residues of linear molecule **1**, Ser-2 to Thr-4 and Val-15 to Lys-17, display random coil-like $\Delta\delta C_{\alpha}H$ values. The strand segments nearer the linker in **1**, Phe-5 to Val-8 and Arg-11 to Leu-14, display $\Delta\delta C_{\alpha}H$ values consistent with β -sheet formation, although each $\Delta\delta C_{\alpha}H$ value is smaller than the corresponding value for **3**. Thus, parallel β -sheet structure is well-developed only for the segments of **1** near the linker, and this region is only partially folded. On the basis of these observations, we regard strand

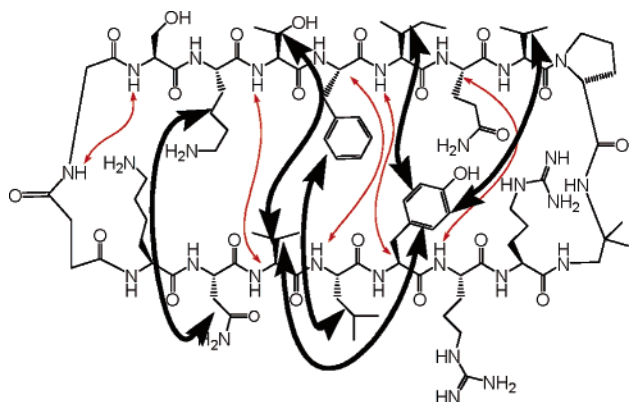


Figure 2. NOEs observed in **3** between residues nonadjacent in sequence. Red arrows indicate backbone-backbone NOEs. Bold arrows indicate multiple NOEs between side chain pairs (at least 3 NOEs for each pair).

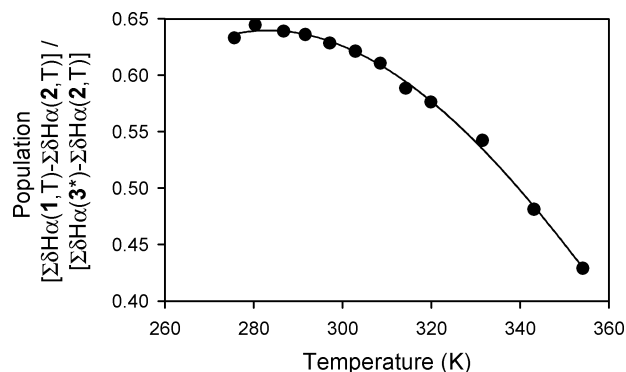


Figure 3. Change in folded population of **1** as a function of temperature, calculated from $\delta C_{\alpha}H$ data by the method of ref 3a. See Supporting Information for details. *Data for **3** at 287 K and 50% TFE.

segments Phe-5 to Val-8 and Arg-11 to Leu-14 as the folded core of **1**, and we focus on this core in the analysis below.

We examined the effect of 2,2,2-trifluoroethanol (TFE) on $\Delta\delta C_{\alpha}H$ for **3** in an effort to determine whether **3** is fully folded in aqueous buffer. Addition of increasing proportions of TFE to aqueous solutions has been shown to induce progressively larger extents of antiparallel β -sheet folding in several designed peptides.¹¹ The $\Delta\delta C_{\alpha}H$ value for each strand residue of **3** became larger as the TFE content was raised from 0% to 30%,⁷ suggesting that the parallel β -sheet conformation is not fully populated in purely aqueous solution. Further increases in TFE proportion to 40% or 50% had relatively little effect,⁷ suggesting that β -sheet population is maximal at 30% TFE. We used $\delta C_{\alpha}H$ values obtained with **3** in 50% TFE to represent the fully folded state in our population analysis of **1**. Parallel β -sheet population at a particular residue of **1** in aqueous solution at a given temperature was estimated by interpolating the observed $\delta C_{\alpha}H$ value between the $\delta C_{\alpha}H$ value for the corresponding residue in unfolded reference compound **2** under the same conditions and the $\delta C_{\alpha}H$ for folded reference compound **3** at 287 K in 50% TFE. For each of the eight residues in the parallel β -sheet core of **1** we compared folded populations determined at 287 and 354 K in aqueous buffer.⁷ The apparent population change is reasonably consistent across this set of residues, which suggests that the eight-residue core can be analyzed in terms of a two-state model, unfolded vs parallel β -sheet. We used the nonlinear fitting method developed by Searle et al.^{3a} for

van't Hoff analysis of two-state folding (Figure 3), which provided the following thermodynamic parameters for parallel β -sheet formation at 298 K: $\Delta H^{\circ} = -1.1 \pm 0.1$ kcal/mol, $\Delta S^{\circ} = -2.5 \pm 0.2$ cal/mol K, $\Delta C_p^{\circ} = -73 \pm 2$ cal/mol K.¹² The thermodynamic signature for parallel β -sheet folding in **1** is qualitatively similar to that observed for a number of antiparallel β -hairpins in that β -sheet formation is enthalpically favorable and entropically unfavorable near room temperature.^{3,13} This signature differs from that of a classical hydrophobic effect, but the observation of a significant and negative heat capacity change upon folding suggests that there is a hydrophobic contribution to the drive for folding, presumably from interstrand side chain-side chain interactions.

The results reported here lay the groundwork for thermodynamic analysis of the factors that control parallel β -sheet folding preferences. Such studies should provide fundamental insight on a structural motif that is very common in proteins and in protein aggregates associated with human diseases.¹⁴

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Supporting Information Available: Information concerning the synthesis of **3**; chemical shift and NOE information on **1**, **2**, and **3**; TFE titration data for **3** and the van't Hoff analysis of folding. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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